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Atomic Theory Matching - Sec. 1 & 2 & 3 ____

1. This device gave scientists their first image of individual atoms.
A. Democritus.
B. John Dalton
C. J.J. Thomson
D. James Chadwick.
E. ...

2. He confirmed the existence of the neutron.
A. Democritus.
B. John Dalton
C. J.J. Thomson
D. James Chadwick.
E. ...

3. He developed the beginnings of the modern atomic theory in the 1800s.
A. Democritus.
B. John Dalton
C. J.J. Thomson
D. James Chadwick.
E. ...

Worksheet - Chapter 5 - Atomic Structure

5 ATOMIC STRUCTURE In lesson 3, you have studied about atoms and molecules as the constituents of matter. You have also learnt that the atoms are the smallest constituents of matter. In lesson 4 you studied about the chemical reactions, their types and the ways to represent them.

5 ATOMIC STRUCTURE - National

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CHAPTER 5, Atomic Structure and the Periodic Table(continued) An atom of neon-22 has two more neutrons in its nucleus than an atom of neon-20.
isotopes true b c a carbon-12; 12 amu
false The values of atomic masses measured in grams are inconveniently small and impractical to work with.

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CHAPTER 5, Atomic Structure and the Periodic Table(continued) Just How Small Is an Atom? (page 108) 6. Suppose you could grind a sample of the element copper into smaller and smaller particles. The smallest particle that could no longer be divided, yet still has the properties of copper, is _____. 7.

5 ATOMIC STRUCTURE AND THE PERIODIC TABLE

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