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Bases. 1. Acids were

recognized primarily

from their sour taste.

Bases were recognized

from their bitter taste

and slippery feel on

skin. 2. In the

Arrhenius definition, an

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acid is a substance that produces hydrogen ions (H^+) when dissolved in water, whereas a base is a substance that produces hydroxide ions (OH^-) in aqueous solution.

Chapter 16 Acids and Bases

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Chemistry. 1. Lime (calcium hydroxide) is the fifth most produced chemical in the United States. What are some uses for lime?

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explained in Chapter 7,
we can balance the

equation. $C_3H_8(g) + 5O_2(g) \rightarrow 3CO_2(g) + 4H_2O(g)$

Check: 3 C, 8 H, 10 O \rightarrow 3 C, 8 H, 10 O

This equation can be interpreted in terms of molecules as follows: 1 molecule of C_3H_8 , 5 molecules of O_2 , 3 molecules of CO_2 , 4 molecules of H_2O

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12. 1 mol O = 16.00 g

O = 6.02×10^{23} O atoms

1 O atom 16.00 g O

6.022×10^{23} O atoms

= 2.66 10^{-23} g O

13. 0.50 mol O atoms

16.00 g O 1 mol = 8.0

g O 4 mol H atoms

1.008 g 1 mol = 4 g H

Half a mole of O atoms

weighs more than 4

moles of H atoms.

Chapter 6 Chemical Composition

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